



Bioorganic & Medicinal Chemistry

Bioorganic & Medicinal Chemistry 13 (2005) 3249-3261

Design, synthesis and anti-plasmodial evaluation in vitro of new 4-aminoquinoline isatin derivatives

Idan Chiyanzu,^a Cailean Clarkson,^b Peter J. Smith,^b Julie Lehman,^c Jiri Gut,^c Philip J. Rosenthal^c and Kelly Chibale^{a,*}

^aDepartment of Chemistry, University of Cape Town, Rondebosch, 7701, South Africa
^bDivision of Pharmacology, Department of Medicine, University of Cape Town, K-45 OMB, Groote Schuur Hospital,
Observatory, 7925, South Africa
^cDepartment of Medicine, San Francisco General Hospital, University of California, San Francisco, Box 0811,

San Francisco, CA 94143, USA

Received 26 January 2005; revised 16 February 2005; accepted 16 February 2005 Available online 19 March 2005

Abstract—A new class of 4-aminoquinoline derivatives based on the natural product isatin scaffold were designed and synthesized for biological evaluation against three strains of the malaria parasite *Plasmodium falciparum*. These derivatives showed anti-plasmodial IC₅₀ values in the ranges of 1.3–0.079 and 2.0–0.050 μM against a chloroquine-sensitive (D10) and two resistant (K1 and W2) strains of *P. falciparum*, respectively. In order to determine potential targets for this class of compounds in *P. falciparum*, selected compounds were also tested against the parasitic cysteine protease falcipain-2. In terms of further development of this class of isatin derivatives, two of the compounds based on a flexible alkyl chain linker and a thiosemicarbazone moiety warrant further investigation as potential anti-plasmodial leads. These two derivatives showed good in vitro activity against K1 and W2 with IC₅₀ values of 51 and 54 nM, respectively, while retaining potency against the D10 strain with IC₅₀ values of 79 and 95 nM, respectively. Generally speaking, the inhibitory potency of all compounds in the series against the parasites did not strongly correlate with inhibitory potency against falcipain-2 for selected compounds tested, which at best was weak to moderate, suggesting other mechanisms of inhibition may also be involved or compounds may be selectively taken up by *Plasmodium falciparum*.

© 2005 Elsevier Ltd. All rights reserved.

1. Introduction

Plasmodium falciparum malaria continues to be a global health problem, especially in developing countries such as sub-Saharan Africa. It is responsible for over 1 million deaths per year. The rapid spread of resistance to available anti-malarial drugs, especially chloroquine and related quinoline-based agents, has highlighted the need to identify alternative anti-malarial compounds. Recent approaches aimed at decreasing and/or slowing down the potential for quinoline-based anti-malarial drug resistance has included the design and synthesis of quinoline-containing dual inhibitors or 'double drugs' that would potentially inhibit haemozoin formation and

another target within *P. falciparum*. This approach is exemplified by cysteine protease inhibitors based on mefloquine and chloroquine² as well as quinoline-based inhibitors of a neutral zinc aminopeptidase,³ primaquine-based plasmepsin inhibitors⁴ and inhibitors of antioxidant enzymes, such as glutathione reductase, conjugated to a 4-aminoquinoline moiety.^{5,6}

Amongst other objectives, our current medicinal chemistry programme is geared towards addressing the problem of drug resistance through the development of single agents that provide maximal anti-infective and/or anti-cancer activity by acting against multiple targets within the same disease-causing organism and/or cell. Within this context, one of the approaches we have taken is to hybridize electrophilic moieties with known minimum structural requirements for biological activity, bioactiphores, from various anti-infective and/or anti-cancer agents. The goal of this approach is to target, amongst other things, cysteine residues of biomolecules by exploiting the strong nucleophilic character of the

Keywords: Isatins; Aminoquinolines; Plasmodium falciparum; Malaria; Anti-plasmodial activity; Falcipain-2; Mannich bases; Thiosemicarbazones.

^{*}Corresponding author. Tel.: +27 21 650 2553; fax: +27 21 689 7499; e-mail: chibale@science.uct.ac.za

thiol (SH) group, as a thiolate, which allows S-alkylation/arylation and S-acylation/sulfonylation of cysteine residues. Indeed a vast number of cysteine protease inhibitors exploit this property of sulfur.^{8–10}

In our hybridization approaches, we have embarked on the rational selection of a variety of electrophilic warheads and their chemical linkage to bioactiphores from a diverse range of anti-infective (anti-malarial, antituberculosis, anti-trypanosomal and anti-HIV/AIDS) and anti-cancer agents. This paper describes the design and synthesis of 4-aminoquinoline-based isatin derivatives 1-4 (Fig. 1), in which the ketone and thiosemicarbazone moieties serve as electrophilic warheads, and reports on their in vitro anti-plasmodial activity against a chloroquine sensitive (D10) and two resistant (K1 and W2) strains of *P. falciparum*. A few non-quinoline Mannich bases 5 were also designed with a view to establishing preliminary structure–activity relationships with respect to the role of the 4-aminoquinoline moiety in the biological activity. In order to determine potential targets for this novel class of aminoquinolines, and in view of the fact that we have already reported on the inhibition by isatins of parasitic cysteine proteases from multiple protozoan parasites, 11 a few selected compounds were screened against a P. falciparum-derived cysteine protease, falcipain-2, a known target for antimalarial agents that block haemoglobin degradation.¹²

The derivatives, 1–4, were designed on the basis of a multi-therapeutic strategy exemplified for quinoline eth-

ylene isatin derivatives **4** in Figure 2. It is postulated that quinoline anti-malarials accumulate in the parasite's acidic food vacuole and inhibit β-haematin formation. Toxic haem thus builds up in the vacuole subsequently killing the parasites. Though the mode of action of these agents is still unclear, they remain potential sources of anti-malarial therapy. This is partly because *P. falciparum* has had difficulty developing rapid resistance to this class of compounds. For example simple modification of the lateral side chain of chloroquine has resulted in new derivatives, including metallocenes, ^{14,15} with activity against chloroquine-resistant strains. ^{13,16,17}

While the isatin unit is a privileged natural product scaffold onto which other bioactiphores can be appended, the thiosemicarbazone moiety has been selected to provide reactive sites (the imine and thiol carbonyl) for alkylation of the enzyme cysteine thiolate. 8,18 This moiety is also likely a metal chelator in which the hydrazinic nitrogen and sulfur atoms could be involved in binding to endogenous iron within P. falciparum. This has implications for metabolism within the parasite by, for example, inhibiting metal-dependent enzymes. The hydrazinic nitrogen could also assist in the accumulation within the acidic food vacuole of P. falciparum. The basic protonatable piperazine nitrogen of the Mannich bases 1 and 2 was envisaged to further increase accumulation of the molecule. The 7-chloroquinoline moiety has been proposed from earlier findings to bind to haematin in the parasite's acidic food vacuole, thus inhibiting

$$R^{1} \xrightarrow{X} O \qquad R^{1} \xrightarrow{X} O \qquad N-NH$$

$$1; X = O$$

$$2; X = NNHC(S)NH_{2}$$

$$3; X = O$$

$$4; X = NNHC(S)NH_{2}$$

Figure 1. Chemical structures of aminoquinoline and Mannich base isatin derivatives 1-5.

Figure 2. Multi-therapeutic strategy for quinoline-ethylene isatin derivatives 4.

haemazoin formation, and also increases accumulation of the drug due to the protonatable quinoline nitrogen. ¹⁹ The variable alkyl chain in 3 and 4 is important not only for circumversion of the chloroquine resistance mechanism but also for lipophilicity, which is an important parameter in the effectiveness of iron chelating antimalarial agents. ²⁰

1.1. Synthesis of target compounds

Chloro-4-piperazin-1-yl-quinoline **8**, the precursor compound for Mannich bases was firstly synthesized in good yield (67%) from condensation of 4,7-dichloroquinoline **6** and piperazine **7** (excess) in *N*-methyl-2-pyrrolidine (NMP), to give the target compound **8** (Scheme 1).

Upon the preparation of the required secondary amine precursor 8, we proceeded with the Mannich reaction. Mannich reactions can be performed via a one pot direct multi-component protocol requiring isatin, an aldehyde and an amine or via a preformed iminium ion. The latter approach was used in this work. Accordingly, paraformaldehyde (1.0 equiv) and chloro-4-piperazin-1-yl-quinoline 8 (1.0 equiv) were dissolved in ethanol and stirred for 30 min. The iminium ion formed in situ was then reacted with selected isatins 9 (Scheme 2a) and isatin-3thiosemicarbazones (Scheme 2b) in ethanol for 3 h at room temperature to furnish the desired N-Mannich isatin derivatives accomplished in low to excellent yields 26–93% (Table 1). The above mentioned approach was used in the preparation of non-quinoline based Mannich derivatives 5.

The quinoline–ethylene isatin derivatives **3** and **4** listed in Table 2 were prepared according to Scheme 3. Reaction of excess 2-amino ethanol **11** and 4,7-dichloroquinoline **6** as reported previously¹⁹ gave **12** in a good yield of 92%. This was followed by *o*-mesylation²¹ in pyridine at 0 °C for 5 h to furnish mesylate **13** in a yield of 82%. Chemoselectivity in the mesylation of preformed alcohol

Scheme 1. Synthesis of chloro-4-piperazin-1yl quinoline 8. Reagents and conditions: (a) K_2CO_3 , Et_3N , NMP, 135 °C, 4 h, 67%.

12 was not a problem in spite of the presence of the 4-amino (NH) group, which is poorly nucleophilic due to the conjugation of the lone pair of electrons into the quinoline nitrogen. The synthesis of quinoline–ethylene isatin derivatives (Scheme 3) employed commercially available isatins and sodium hydride as the base in *N*,*N*-dimethylformamide (DMF) as the solvent resulting in good yields (68–88%) of the products (Table 2). Selected precursors were used in the synthesis of thiosemicarbazones, (Scheme 3).

1.2. Biological evaluation

IC₅₀ values against recombinant falcipain-2 were determined essentially as described previously. Accordingly, equal amounts (\sim 1 nM) each of recombinant protein were incubated with different concentrations of inhibitors added from 100× stocks in DMSO in 100 mM sodium acetate (pH 5.5), -10 mM dithiothreitol for 30 min at room temperature before addition of the substrate benzoxycarbonyl-Leu-Arg-7-amino-4-methyl-coumarin (final concentration, 25 μ M). Fluorescence was continuously monitored for 30 min at room temperature in a Labsystems Fluoroskan II spectrofluorometer. IC₅₀ values were determined from plots of activity over inhibitor concentration with GraphPad Prism software.

Parasites were cultured (D10 and K1) essentially as described by Trager and Jensen.²³ Initial 1 mg/mL stocks of the compounds were made up in DMSO, diluted with water and finally complete medium on the day of the experiment. The highest concentration of DMSO that the parasites were exposed to was 0.05%, which had no measurable effect on parasite viability. No attempt was made to determine IC₅₀ values greater than 1000 ng/mL. All experiments were performed in duplicate on a single occasion using a chloroquine-sensitive strain of P. falciparum (D10). The more active compounds were also tested against a chloroquine-resistant strain of P. falciparum (K1). After culture with inhibitors for 48 h, parasite lactate dehydrogenase (pLDH) activity was used to measure parasite viability as described by Makler et al.24

Ring stage *P. falciparum* strain W2 parasites were synchronized for 15 min at 37 °C 5% sorbitol, adjusted to 2% haematocrit 1% parasitaemia, and incubated at 37 °C under the atmosphere of 3% O₂, 5% CO₂, 91% N₂ in mediumRPMI-1640 supplemented with 10% human serum in the presence of various concentrations of inhibitors for 48 h without media change. Inhibitors

$$R^1$$
 R^1
 R^2
 R^3
 R^3
 R^3

Scheme 2. Synthesis of Mannich bases 1, 2 and 5. Reagents and conditions: (a) CH₂O, 8, EtOH, rt, 3 h, 26–93%; (b) CH₂O, secondary amine, EtOH, 45 °C, 3 h, 26–93%.

Table 1. Chemical and physical data of quinoline and non-quinoline Mannich based isatin derivatives

Compound	\mathbb{R}^1	\mathbb{R}^2	$(R^3)R^4$	X	Time (h)	% Yield	Mp (°C)
1a	Н	Н	_	0	4	93	203–206
1b	Me	Me	_	O	4	85	209-213
1c	C1	H	_	O	4	24	153-157
1d	F	H	_	0	4	94	110-117
1e	Me	H	_	0	4	55	202-203
1f	I	H	_	0	4	59	151-153
1g	Br	H	_	0	4	48	210-212
2a	H	H	_	NNHC(S)NH ₂	3	78	216-217
2b	F	H	_	NNHC(S)NH ₂	3	51	233-234
2c	Cl	H	_	NNHC(S)NH ₂	3	86	234-235
2d	Br	H	_	NNHC(S)NH ₂	3	73	236-237
2e	Me	H	_	NNHC(S)NH ₂	3	78	232-234
5a	H	H	NEt_2	NNHC(S)NH ₂	3	54	134-135
5b	Н	Н	N	NNHC(S)NH ₂	3	95	179–181
5c	Н	Н	N	NNHC(S)NH ₂	3	90	181–184
5d	Н	Н	NO	NNHC(S)NH ₂	3	92	208-210
5e	Н	Н	N_N_Ph	NNHC(S)NH ₂	3	62	187–188
5f	Н	Н	N_N-Ph	NNHC(S)NH ₂	3	76	188–191

Scheme 3. Synthesis of quinoline–ethylene isatin derivatives 3 and 4. Reagents and conditions: (a) 0.3 equiv of K_2CO_3 , 0.3 equiv of Et_3N , reflux, 8 h, 92%; (b) 2.5 equiv of methanesulfonyl chloride, pyridine, 0 °C, 5 h, 83%; (c) NaH, isatin/5-substituted isatin, DMF, 60 °C, 24 h, 68–89%; (d) $H_2NNHC(S)NH_2$, ethanol, 3 h, 45 °C, 68–89%.

were added from DMSO dissolved stocks, maximum concentration of DMSO used was 0.1%. Two independent assays are performed in 200 μL final volume in flat bottom 96-well plates.

After 48 h the culture medium was removed and replaced with 1% formaldehyde in PBS pH 7.4, and fixed for 48 h at room temperature. Fixed parasites were transferred into 0.1% Triton-X-100 in PBS containing

Table 2. Chemical and physical data of a series of ethylene-quinoline isatin derivatives

Compound	\mathbb{R}^1	\mathbb{R}^2	X	Time (h)	%Yield	Mp (°C)
3a	Н	Н	0	16	88	240–243
3b	Me	H	O	16	68	240-243
3c	C1	Н	O	16	69	221-223
3d	Me	Me	O	16	89	219
3e	I	H	O	16	70	200-201
4a	Н	Н	NNHC(S)NH ₂	3	70	167-171
4b	Me	Н	NNHC(S)NH ₂	3	89	227-230
4c	C1	Н	NNHC(S)NH ₂	3	68	221-227

1 nM dye YOYO-1 (Molecular Probes). Parasitaemia was determined from dot plots (forward scatter vs Fluorescence FL-1) acquired on FACSort flow cytometer using CellQuest software (Beckton Dickinson).²⁵

The 50% inhibitory concentration (IC $_{50}$) values for the compounds were obtained from the dose–response curves, using non-linear dose–response curve fitting analyses with GraphPad Prism v.3.00 software.

2. Results

The results of the biological evaluations are presented in Table 3. Quinoline thiosemicarbazone derivatives 2 and

4 generally showed better inhibition of falcipain-2 compared to the corresponding ketones 1 and 3. It is noteworthy that all non-quinoline Mannich base thiosemicarbazones 5 were less active compared to the corresponding quinoline Mannich bases 4 (IC₅₀ > 20 μ M). This may suggest preference for larger hydrophobic groups in the S2 enzyme binding pocket.

Within the thiosemicarbazones, quinoline-Mannich base derivatives ${\bf 1a}$, ${\bf 1c}$ and ${\bf 1e}$ were more active against falcipain-2, with IC₅₀ values mostly less than 10 μ M, compared to the ethylene quinolines ${\bf 4a}$ – ${\bf b}$. Considering compounds that were similar (${\bf 2a}$ – ${\bf d}$) with respect to position 5 substituents, the general order of activity against falcipain-2 was Br > F > H. This trend is consistent with

Table 3. IC₅₀ values of compounds against chloroquine sensitive (D10), chloroquine resistant (K1 and W2) strains and falcipain-2

Compound	\mathbb{R}^1	\mathbb{R}^2	X	IC_{50} , μM				$\operatorname{Clog} P$
				D10 ^a	K1 ^b	W2 ^b	FP-2 ^c	
1a	Н	Н	0	0.95	ND^d	0.96	>20	3.31
1c	Cl	Н	O	1.07	ND	ND	ND	4.32
1d	F	H	O	0.72	0.72	0.91	>20	3.75
1f	I	H	O	0.92	ND	0.79	20	4.61
2a	H	Н	NNHC(S)NH ₂	1.37	ND	0.96	9.26	4.49
2b	F	H	NNHC(S)NH ₂	1.18	ND	1.19	7.99	4.63
2c	Cl	Н	NNHC(S)NH ₂	1.37	ND	1.18	ND	5.20
2d	Br	H	NNHC(S)NH ₂	0.64	0.52	1.23	6.07	5.35
3a	H	Н	O	0.90	0.99	0.12	>20	3.88
3b	Me	H	O	0.55	0.73	0.49	>20	4.38
3c	Cl	H	O	0.38	1.51	ND	ND	4.77
3d	Me	Me	O	0.33	0.45	0.23	20	4.88
4a	H	H	NNHC(S)NH ₂	0.32	0.71	0.24	14.65	5.15
4b	Me	H	NNHC(S)NH ₂	0.079	0.10	0.051	11.64	5.65
4c	Cl	H	NNHC(S)NH ₂	0.095	0.054	ND	ND	5.86
CQ			. , -	0.033	0.312	0.240	ND	5.08

^a Chloroquine-sensitive strain.

^b Chloroquine-resistant strain.

^c Falcipain-2.

^d Not determined.

the trend previously observed for isatin-3-thiosemicarbazones in which a large (iodo) substituent at position 5 resulted in high potency against the enzyme.¹¹

In the study of anti-malarial effects of compounds 1–4, it is evident that thiosemicarbazones 4a, 4b and 4c were generally superior compared to the corresponding ketone derivatives 3a, 3b and 3c. This trend was, however, not observed for the Mannich bases 1 and 2. Compounds 4b and 4c were particularly active against all three P. falciparum strains. These two compounds were even superior to chloroquine in the resistant W2 and K1 strains, with IC₅₀ values of 0.051 and 0.054 µM, respectively, while still retaining respectable activity against D10. Within the thiosemicarbazones, the ethylene-based derivatives were comparatively more potent than the Mannich bases. For example, compounds 4b and 4c had better IC₅₀ values, roughly all less than 1 μM. Removal of the quinoline ring system generally resulted in a significant loss of activity against all the three parasite strains (compounds 5a-f; IC₅₀ values >10 μ M for strain W2). Among the more active ethylene–quinoline thiosemicarbazones, methyl substitution at position 5 on the isatin ring resulted in better inhibitors of parasite growth in W2 strain. Compound 4b, with $IC_{50} =$ 0.051 µM, was superior (5-fold) to chloroquine in this strain. Chloro substitution at position 5, as in 4c, also showed excellent inhibitory activity against K1 with an IC_{50} value of 0.054 μ M, which was also superior (6-fold) to chloroquine. Considering compounds of series 4 only, the order of anti-malarial activity in K1 was generally Cl > Me > H. In all cases, the biological assays did not show the food vacuole abnormality that accompanies intracellular falcipain-2 inhibition, that is, swelling due to accumulation of undigested haemoglobin. The SAR study within this class of compounds clearly shows no strong correlation between inhibition of falcipain-2 and anti-malarial activity.

3. Discussion

We have shown that aminoquinoline thiosemicarbazone derivatives of isatin are potent inhibitors of P. falciparum growth. Considering inhibition of the protease falcipain-2, quinoline-Mannich base isatin derivatives (2a, 2b and 2d) blocked activity at IC₅₀ values of less than 10 μM. Compounds 4a, 4b and 4c (quinoline-ethylene isatin derivatives) effectively blocked parasite development, but had modest activity against falcipain-2, suggesting a lack of strong correlation between the potencies of compounds against falcipain-2 and parasites. This suggests that the anti-malarial activity of inhibitors 4a, 4b and 4c were not entirely due to inhibition of falcipain-2 and that other mechanisms, including selective uptake of the compounds by the parasite, are involved. Although falcipain-2 appears not to be the principal target for these compounds, it is possible that the compounds may be acting through the inhibition of another cysteine protease since there are many cathepsin L-like cysteine proteases in P. falciparum. $^{25-27}$ The possibility that these compounds may be exerting their antimalarial effects through inhibition of another cysteine

protease in *P. falciparum* is supported by the fact that isatins such as compounds **2a**, **2b**, **2d**, **4a** and **4b** can bind to and inhibit a cysteine protease. The inhibitory activity (albeit modest) as demonstrated against falcipain-2 is a case in point. Alternatively compounds such as **4a**, **4b**, **4c** may be acting via a protease-independent mechanism. Similar conclusions have recently been made for tridentate chelating 2-acetylpyridine thiosemicarbazones. Our results further suggest that compounds that share structural features of the potent compounds that we have identified, but also have improved intracellular inhibition of falcipain-2, should offer excellent antimalarial activity.

The excellent inhibitory results against parasite development by compounds 4a, 4b and 4c led to the speculation that improved potency could have been due to the contribution from specific moieties notably thiosemicarbazone and 4-aminoquinoline moieities, the ethylene linker and substituents at position 5 of the isatin scaffold (chloro and methyl substitution at position 5 on the aromatic ring of the isatin is favoured presumably for binding to the target). Thiosemicarbazones are potential metal-chelators in P. falciparum, which may exhibit biological activity by a number of mechanisms including: (a) metal-interactive inhibition of cysteine proteases; ²⁹ (b) inhibition of metal-dependent enzymes such as falcilysin,³⁰ ribonucleotide reductase^{31,32} and aminopeptidases;3 (c) formation of lethal complexes, which are directly toxic to the parasite as a result of the induction of oxidative stress.²⁸

Introduction of the ethylene linker in the molecule was generally more favoured compared to Mannich bases with a piperazinyl linker. The ethylene linker increases the lipophilicity of compounds 4a, 4b, 4c compared to 1 and 2. For example compounds 4a and 4c have higher Clog P values compared to the corresponding Mannich base compounds 2a and 2c, which probably aids their passage through parasite membranes on the way to their presumed site of action, the acidic food vacuole. If these compounds are also exerting their anti-malarial effects via chelation of iron in *P. falciparum*, then high lipophilicity is a major contributor. Lipophilicity is an important physical property of iron chelators since the iron withheld by chelators from *P. falciparum* most likely resides within the parasitic compartment of the malaria infected red blood cell. As such an effective anti-malarial iron chelator should have the ability to cross-lipid membranes and display high selective affinity for iron (II) or iron (III) compared to other endogenous transition metals. A correlation between the degree of lipophilicity of an iron chelator and its anti-plasmodial activity has been demonstrated.²⁰

4. Conclusion

In summary we have designed compounds based on a multi-therapeutic strategy that were effective against three strains of the malaria parasite *P. falciparum* and with potential to inhibit *P. falciparum*-derived cysteine proteases. Two compounds were particularly effective

against chloroquine-resistant strains, thus validating their ability to evade the chloroquine resistance mechanism and cross-membranes to reach their site of action. These results indicate that multi-therapeutic strategies are promising strategies for identifying hits for subsequent optimization into potential leads against drug resistant parasites. Given the presumed multiple targets for the different moieties of these compounds, each mode of action needs to be determined with a view to establishing the mechanisms(s) most responsible for the observed anti-malarial activity.

5. Experimental

5.1. General

Proton nuclear magnetic resonance (¹H NMR) spectra were recorded at ambient temperatures using the following instruments: Varian Mercury (300 MHz) or Varian Unity Spectrometer (400 MHz) and TMS was used as an internal standard. The chemical shifts (δ) are given in parts per million relative to TMS ($\delta = 0.00$). Carbon-13 nuclear magnetic resonance (¹³C NMR) spectra were recorded at 75 MHz or 100 MHz with the same internal standard. Diverse solvents were used in the determination of spectra for different compounds. The chemical shifts (δ) are given in ppm relative to TMS $(\delta = 0.00)$. Mass spectra were recorded by means of a VG micromass 16 F spectrometer at 70 eV with accelerating voltage 4 kV. Accurate masses were determined using a VG-70E spectrometer and VG (Micromass) 70-SE magnetic sector mass spectrometer. Infrared spectra were measured either in solution form using chloroform or as solids (KBr pellets) or Nujol mulls on a satellite FT-IR spectrophotometer. Micro (elemental) analysis was performed using a Fisons EA 1108 CHNS-O instrument. Melting points were determined by using a Reicher–Jung Thermovar (temperature range 0-350 °C) on cover slips and are uncorrected. Column chromatography and preparative layer chromatography (p.l.c), carried out on silica gel (Merck Kieselgel 60) were used in purification of samples. Reactions were monitored by thin-layer chromatography (TLC) using coated silica gel plates, detection by an ultra-violet lamp. The lipophilicity of the compounds was estimated via the Clog P values, which were calculated using Chem-Draw Ultra 6.0.

5.1.1. 7-Chloro-4-piperazin-1-yl-quinoline (8). A mixture of 4,7-dichloroquinoline (2.0 g, 10.10 mmol), piperazine (4.35 g, 50.49 mmol), K_2CO_3 (0.419 g, 3.03 mmol), triethylamine (0.423 mL, 3.02 mmol) in *N*-methyl-2-pyrrolidinone (10 mL) was stirred under nitrogen at 135 °C for 4 h. After cooling to room temperature, the mixture was diluted with CH_2Cl_2 (50 mL) and washed with brine (10 × 50 mL). The organic layer was separated, dried over sodium sulfate, filtered and concentrated. Column chromatography (SiO₂; MeOH/EtOAc, 2:80) yielded the amine **8** (1.87 g, 67%) as a white solid; mp 113–114 °C (from methanol); R_f 0.34 (MeOH/EtOAc, 2:8); δ_H (300 MHz; CDCl₃); 8.70 (1H, d, *J* 5.1, H-2'), 8.02 (1H, d, *J* 2.1, H-8'), 7.93 (1H, d, *J* 9.0, H-5'), 7.40

(1H, dd, J 2.1 and 9.0, H-6'), 6.81 (1H, d, J 5.1, H-3'), 3.17–3.12 (8H, br s, H-2 and H-3); $\delta_{\rm C}$ (75 MHz); 157.3, 151.9, 150.2, 134.8, 128.9, 126.1, 125.2, 121.9, 108.9, 53.5 (2 × C) and 46.1 (2 × C); LRMS (EI) m/z 247 [66%, M⁺] and 205 [100%, M-C₁₁H₁₀N₂CI]. Anal Calcd for C₁₃H₁₄N₃CI: C 63.0%, H 5.7%, N 16.9%. Found: C 62.8%, H 5.6%, N 16.6%.

5.2. A. General procedure for preparing compounds 1a-g

To a solution of isatin or substituted isatin (0.2 g, 1.35 mmol) in 5 mL of 99.9% ethanol was added to a mixture of compound **8** (0.27 g, 1.35 mmol) and aqueous formaldehyde 37% (0.03 mL, 1.35 mmol) also dissolved in 10 mL of 99.9% ethanol. The reaction mixture was stirred for 3 h at room temperature, refrigerated for 48 h to form crystals. The crystalline products were separated by filtration, washed and vacuum dried. Recrystallization from ethanol rendered desired products in pure form.

5.2.1. N^1 -[4-(7-Chloro-quinolin-4-yl)-piperazin-1-ylmeth**vll-1***H***-indole-2,3-dione** (1a). (0.51 g, 93%) obtained as reddish-orange crystals; mp 203–206 °C (from ethanol); IR (KBr) $v_{\rm max}/{\rm cm}^{-1}$ 1730 (keto C=O), 1617 (amide C=O), 1572 (C=N); δ_H (300 MHz; CDCl₃); 8.70 (1H, d, J 5.1, H-2"), 8.02 (1H, d, J 2.1, H-8"), 7.92 (1H, d, J 9.0, H-5"), 7.66 (1H, dd, J, 0.6 and 8.0, H-4), 7.62 (1H, ddd, J 0.6, 6.8 and 7.8, H-6), 7.43 (1H, dd, J 2.1 and 9.0, H-6"), 7.19 (1H, ddd, J 0.9, 6.8 and 8.0 and H-5), 7.10 (1H, d, J 7.8, H-7), 6.82 (1H, d, J 5.1, H-3"), 4.60 (2H, s, $-NCH_2N_-$), 3.24 (4H, br s, $-N(CH_2CH_2)_2NAr$), 2.95 (4H, br s, $-N(CH_2CH_2)_2NAr); \delta_C$ (75 MHz); 182.1, 159.0, 156.6, 151.8, 151.5, 150.1, 138.4, 128.9, 128.4, 126.3, 125.4, 125.2, 124.0, 121.8, 117.4, 111.5, 109.1, 62.3, 51.8 $(2 \times C)$ and 50.6 $(2 \times C)$; HRMS (FAB) found m/z $407.12661 \text{ [M+1]}^+, C_{22}H_{19}O_2N_4C1 \text{ requires } 407.12747.$ Anal Calcd for $C_{22}H_{19}O_2N_4C1$ C 64.4%, H 4.7%, N 13.3%. Found: C 64.6%, H 4.5%, N 13.3%.

5.2.2. N^1 -[4-(7-Chloro-quinolin-4-yl)-piperazin-1-ylmethyl]-5,7-dimethyl-1H-indole-2,3-dione (1b). The conditions employed for the preparation of this compound were those described in general procedure **A**. However, 1.14 mmol of isatin and compound **8** were used.

(0.42 g, 85%) obtained as red crystals; mp 209–213 °C (from ethanol); IR (CHCl₃) $v_{\text{max}}/\text{cm}^{-1}$ 1739 (keto C=O), 1635 (amide C=O), 1542 (C=N); $\delta_{\rm H}$ (400 MHz; CDCl₃); 8.71 (1H, d, J 5.2, H-2"), 8.02 (1H, d, J 2.0, H-8"), 7.93 (1H, d, J 9.0, H-5"), 7.43 (1H, dd, J, 2.0 and 9.0, H-6"), 7.31 (1H, d, J 2.0, H-4), 7.20 (1H, d, J 2.0, H-6), 6.80 (1H, d, J 5.2, H-3"), 4.51 (2H, s, $-NCH_2N_-$), 3.23 (4H, br s, $-N_ CH_2CH_2NAr$), 2.86 (4H, br s, $-NCH_2CH_2NAr$), 2.58 (3H, s, Ar–C H_3), 2.29 (3H, s, Ar–C H_3); δ_C (75 MHz); 182.7, 159.0, 157.7, 151.1, 150.1, 138.9, 134.9, 128.8, 128.7, 126.8, 125.9, 125.6, 125.0, 122.2, 118.4, 111.8, 109.8, 62.4, 52.8 (2 \times C), 50.6 (2 \times C), 20.1 and 19.1; LRMS (EI) m/z 434 [58%, M⁺], 119 [100%, M-C₈H₉N]. Anal. Calcd for C₂₄H₂₃O₂N₄Cl: C 66.3%, H 5.3%, N 12.9%. Found: C 67.3%, H 5.5%, N 12.3%.

5.2.3. N^1 -[4-(7-Chloro-quinolin-4-yl)-piperazin-1-ylmeth-yl]-5-chloro-1H-indole-2,3-dione (1c). The conditions employed for the preparation of this compound were those described in general procedure **A.** However, 1.10 mmol of isatin and compound **8** were used.

(0.13 g, 24%) obtained as orange crystals; mp 153-157 °C (from ethanol); IR $v_{\text{max}}/\text{cm}^{-1}$ 1736 (keto C=O), 1617 (amide C=O), 1511 (C=N); $\delta_{\rm H}$ (300 MHz; CDCl₃); 8.69 (1H, d, J 5.0, H-2"), 8.00 (1H, d, J 2.0, H-8"), 7.89 (1H, d, J 8.8, H-5"), 7.61 (1H, d, J 2.4, H-4), 7.57 (1H, dd, J 2.4 and 8.4, H-6), 7.42 (1H, dd, J 2.0 and 8.8, H-6"), 7.09 (1H, d, J 8.4, H-7), 6.80 (1H, d, J 5.0, H-3"), 4.57 (2H, s, -NCH₂N-), 3.21 (4H, br s, -NCH₂CH₂NAr), 2.91 (4H, br s, $-NCH_2CH_2NAr$); δ_C (75 MHz); 181.9, 158.3, 156.6, 151.7, 149.9, 149.5, 138.8, 128.9, 128.4, 126.3, 125.2, 124.9, 123.7, 121.7, 118.4, 112.9, 109.1, 62.4, 51.8 $(2 \times C)$ and 50.6 $(2 \times C)$; LRMS (EI) m/z 440 [3%, M^{+}], 260 [100%, $M-C_{14}H_{15}ClN_{3}$]. Anal. Calcd for $C_{22}H_{18}O_2N_4Cl_2\cdot H_2O$: C 57.6%, H 4.1%, N 12.2%. Found: C 57.6%, H 4.1%, N 11.9%.

5.2.4. N^1 -[4-(7-Chloro-quinolin-4-yl)-piperazin-1-ylmeth-yl]-5-fluoro-1H-indole-2,3-dione (1d). The conditions employed for the preparation of this compound were those described in general procedure **A**. However, 1.21 mmol of isatin and compound **8** were used.

(0.49 g, 94%) obtained as orange crystals; mp 110-117 °C (from ethanol); IR (CHCl₃) $v_{\text{max}}/\text{cm}^{-1}$. 1744 (keto C=O), 1615 (amide C=O), 1562 (C=N); $\delta_{\rm H}$ (400 MHz; CDCl₃); 8.70 (1H, d, J 5.2, H-2"), 8.03 (1H, d, J 2.0, H-8"), 7.90 (1H, d, J 9.0, H-5"), 7.42 (1H, dd, J 2.0 and 9.0, H-6"), 7.35 (1H, d, J 2.1, H-4), 7.31 (1H, dd, J 2.1 and 8.2, H-6), 7.11 (1H, d, J 8.2, H-7), 6.82 (1H, d, J 5.2, H-3"), 4.58 (2H, s, -NCH₂N-), 3.25 (4H, br s, $-NCH_2CH_2NAr$), 2.93 (4H, br s, $-NCH_2CH_2NAr$); δ_C (75 MHz); 181.9, 159.6, 156.7, 151.8, 150.1, 150.0, 136.7, 136.1, 128.8, 126.7, 125.3, 125.0, 123.6, 121.9, 119.3, 112.4, 109.2, 62.7, 52.1 $(2 \times C)$, 50.8 $(2 \times C)$; HRMS (FAB) m/z 425.11703 $[M+1]^+$ C₂₂H₁₈ClFN₄O₂ requires 425.11805. Anal. Calcd for C₂₂H₁₈ClFN₄O₂·H₂O: C 60.1%, H 4.3%, N 11.8%. Found: C 60.4%, H 4.3%, N 11.5%.

5.2.5. N^1 -[4-(7-Chloro-quinolin-4-yl)-piperazin-1-ylmeth-yl]-5-methyl-1H-indole-2, 3-dione (1e). The conditions employed for the preparation of this compound were those described in general procedure **A**. However, 1.24 mmol of isatin and compound **8** were used.

(0.29 g, 55%) obtained as reddish-orange crystals; mp 202–203 °C (from ethanol); IR (CHCl₃) $v_{\rm max}/{\rm cm}^{-1}$, 1729 (ketone C=O), 1609 (amide C=O), 1518 (C=N); $\delta_{\rm H}$ (300 MHz; CDCl₃); 8.69 (1H, d, J 5.1, H-2"), 8.02 (1H, d, J 2.1, H-8"), 7.90 (1H, d, J 8.7, H-5"), 7.45 (1H, d, J 2.4, H-4), 7.43 (1H, dd, J 2.4 and 7.8, H-6), 7.38 (1H, dd, J 2.1 and 8.7, H-6"), 7.02 (1H, d, J 7.8, H-7), 6.81 (1H, d, J 5.1, H-3"), 4.58 (2H, s, $-{\rm NC}H_2{\rm N}-{\rm NC}H_2{\rm N}-{\rm NC}H_2{\rm CH}_2{\rm NAr}$), 2.94 (4H, br s, $-{\rm NC}H_2{\rm CH}_2{\rm NAr}$), 2.35 (3H, s, Ar–C H_3); $\delta_{\rm C}$ (75 MHz); 182.4, 159.1, 156.6, 151.5, 150.7, 150.2, 138.4, 128.8,

128.4, 126.4, 125.7, 125.6, 124.4, 121.8, 118.2, 110.5, 109.0, 62.3, 51.6 (2 × C), 50.6 (2 × C) and 21.2; HRMS (FAB) found m/z 421.14346 [M+1]⁺ $C_{23}H_{21}ClN_4O_2$ requires 421.14312. Anal. Calcd for $C_{23}H_{21}ClN_4O_2 \cdot H_2O$: C 62.8%, H 4.8%, N 12.8%. Found: C 63.0%, H 4.9%, N 12.9%.

5.2.6. N^1 -[4-(7-Chloro-quinolin-4-yl)-piperazin-1-ylmeth-yl]-5-iodo-1*H*-indole-2, 3-dione (1f). The conditions employed for the preparation of this compound were those described in general procedure **A**. However, 0.73 mmol of isatin and compound **8** were used.

(0.23 g, 59%) obtained as red crystals; mp 151–153 °C (from ethanol); IR (CHCl₃) $v_{\rm max}/{\rm cm}^{-1}$ 1745 (keto C=O), 1602 (amide C=O), 1536 (C=N); $\delta_{\rm H}$ (400 MHz; CDCl₃); 8.68 (1H, d, J 5.2, H-2"), 8.02 (1H, d, J 2.0, H-8"), 7.94 (1H, d, J 9.0, H-5"), 7.93 (1H, d, J 0.3, H-4), 7.90 (1H, dd, J 0.3 and 8.0, H-6), 7.43 (1H, dd, J 2.0 and 9.0, H-6"), 6.94 (1H, d, J 8.0, H-7), 6.81 (1H, d, J 5.2, H-3"), 4.58 (2H, s, -NC H_2 N-), 3.22 (4H, br s, -NCH₂C H_2 NAr); $\delta_{\rm C}$ (75 MHz); 181.6, 158.6, 156.9, 151.8, 150.7, 150.0, 136.7, 135.8, 128.8, 126.4, 124.9, 124.5, 123.9, 121.8, 119.2, 113.7, 109.1, 62.4, 51.8 (2×C) and 50.2 (2×C); HRMS (FAB) found m/z 534.03212 [M+1]⁺ C₂₂H₁₈ClIN₄O₂ requires 534.03195. Anal. Calcd for C₂₂H₁₈ClIN₄O₂ 0.25H₂O: C 49.2%, H 3.4%, N 10.4%. Found: C 49.2%, H 3.1%, N10.3%.

5.2.7. N^1 -[4-(7-Chloro-quinolin-4-yl)-piperazin-1-ylmeth-yl]-5-bromo-1H-indole-2, 3-dione (1g). The conditions employed for the preparation of this compound were those described in general procedure **A**. However, 0.88 mmol of isatin and compound **8** were used.

(0.21 g, 48%) obtained as orange crystals; mp 210-212 °C (from ethanol); IR (CHCl₃) v_{max}/cm^{-1} (keto C=O), 1618 (amide C=O), 1502 (C=N); $\delta_{\rm H}$ (300 MHz; CDCl₃); 8.70 (1H, d, J 5.1, H-2"), 8.02 (1H, d, J 2.1, H-8"), 7.90 (1H, d, J 9.0, H-5"), 7.72 (1H, d, J 2.1, H-4), 7.55 (1H, dd, J 2.1 and 8.4, H-6), 7.42 (1H, dd, J 2.1 and 9.0, H-6"), 7.05 (1H, d, J 8.4, H-7), 6.82 (1H, d, J 5.2, H-3"), 4.58 (2H, s, -NCH₂N-), 3.24 (4H, br s, -NCH₂CH₂NAr), 2.93 (4H, br s, $-NCH_2CH_2NAr$); δ_C (75 MHz); 182.2, 161.1, 156.6, 151.8, 150.8, 150.0, 136.1, 128.8, 128.4, 126.4, 126.0, 124.9, 124.0, 121.8, 118.1, 112.5, 109.0, 62.4, 51.8 $(2 \times C)$ and 50.6 $(2 \times C)$; LRMS (EI) m/z 486 [3%, M^{+}], 260 [100%, $M-C_{14}H_{15}ClN_{3}$]. Anal. Calcd for C₂₂H₁₈O₂BrN₄Cl·H₂O: C 52.4%, H 3.6%, N 11.5%. Found: C 52.7%, H 3.5%, N 11.6%.

5.3. B. General procedure for the preparation of compounds 2a-e

To a solution of isatin or substituted isatin-3-thiosemicarbazone derivative (0.2 g, 1.36 mmol) in 5 mL of 99.9% ethanol was added a mixture of compound **8** (0.22 g, 1.36 mmol) and aqueous formaldehyde 37% (0.027 mL, 1.35 mmol) also dissolved in 10 mL ethanol (99.9%). The reaction mixture was stirred for 3 h at room temperature for isatins or 45 °C for isatin-3-thio-

semicarbazones, refrigerated for 48 h to form crystals. The crystalline product was separated by filtration, vacuum dried and recrystallized from ethanol.

5.3.1. N^1 -[4-(7-Chloro-quinolin-4-yl)-piperazin-1-ylmethyl-1*H*-indole-2,3-dione 3-thiosemicarbazone (1a). (0.29 g, 78%) obtained as orange crystals; mp 216–217 °C (from ethanol); IR (KBr): $v_{\text{max}}/\text{cm}^{-1}$ 3686 (N–H), 3036 (Ar–H), 1675 (C=O), 1511 (C=N), 1122 (C=S); δ_{H} (300 MHz; DMSO-d₆); 12.40 (1H, s, NNH), 9.04 (1H, s, NHH), 8.69 (1H, s, NHH), 8.65 (1H, d, J 5.2, H-2"), 7.96 (1H, d, J 9.0, H-5"), 7.91 (1H, d, J 2.4, H-8"), 7.66 (1H, dd, J 1.2 and 7.6, H-4), 7.48 (1H, dd, J 2.4 and 9.0, H-6"), 7.42 (1H, ddd, J 1.2, 7.2 and 8.0, H-6), 7.32 (1H, d, J 7.2, H-7), 7.15 (1H, ddd, J 0.4, 7.6 and 8.0, H-5), 6.61 (1H, d, J 5.2, H-3"), 4.60 (2H, s, $-NCH_2N_{-}$), 3.13 (4H, br s, $-NCH_2CH_2NAr$), 2.87 (4H, br s, $-NCH_2CH_2NAr$); δ_C (75 MHz); 179.4, 162.6, 156.9, 152.8, 150.3, 144.5, 134.2, 131.8, 131.7, 128.9, 126.8, 126.4, 123.1, 122.1, 121.3, 120.1, 111.7, 110.2, 61.7, 52.3 (2×C) and 50.7 (2×C); HRMS (FAB) m/z 480.13677 [M+1]⁺, C₂₃H₂₂ClN₇OS requires 480.13732. Anal. Calcd for C₂₃H₂₂ClO₅N₇: C 57.6%, H 4.6%, N 20.4%, S 6.7%. Found: C 57.6%, H 4.7%, N 20.4%, S 6.8%.

5.3.2. N^1 -[4-(7-Chloro-quinolin-4-yl)-piperazin-1-ylmeth-yl-5-chloro-1H-indole-2,3-dione 3-thiosemicarbazone (2c). The conditions employed for the preparation of this compound were those described in general procedure **B**. However, 0.37 mmol of compound 1c and thiosemicarbazide were used.

(0.35 g, 86%) obtained as orange crystals; mp 234-235 °C (from ethanol); IR (KBr): $v_{\text{max}}/\text{cm}^{-1}$ 3687 (N– H), 1675 (C=O), 1518 (C=N), 1120 (C=S); δ_{H} (300 MHz; CDCl₃); 12.3 (1H, s, NNH), 9.10 (1H, s, NHH), 8.71 (1H, s, NHH), 8.68 (1H, d, J 5.4, H-2"), 8.01 (1H, d, J 2.0, H-8"), 7.88 (1H, d, J 8.8, H-5"), 7.61 (1H, d, J 2.1, H-4), 7.59 (1H, dd, J 2.1 and 8.4, H-6), 7.42 (1H, dd, J 2.0 and 8.8, H-6"), 7.11 (1H, d, J 8.4, H-7), 6.81 (1H, d, J 5.4, H-3"), 4.58 (2H, s, $-NCH_2N_{-}$), 3.21 (4H, br s, $-NCH_2CH_2NAr$), 2.91 (4H, br s, $-NCH_2CH_2NAr$); δ_C (75 MHz); 178.7, 162.6, 158.3, 156.7, 151.7, 150.1, 149.5, 138.9, 128.9, 128.4, 126.4, 125.2,123.9, 123.7, 121.7, 119.1, 111.9, 110.0, 62.4, 51.9 (2 \times C), 51.2 (2 \times C). Anal. Calcd for C₂₃H₂₁Cl₂N₇OS: C 53.8%, H 4.1%, N 18.6% S 5.8%. Found: C 53.7%, H 4.1%, N 19.0%, S 6.2%; HRMS (FAB) m/z 514.09889 [M+1]⁺ C₂₃H₂₁Cl₂N₇OS requires 514.09836.

5.3.3. N^1 -[4-(7-Chloro-quinolin-4-yl)-piperazin-1-ylmeth-yl-5-bromo-1H-indole-2,3-dione3-thiosemicarbazone (2d). The conditions employed for the preparation of this compound were those described in general procedure **B.** However, 0.67 mmol of compound 1d and thiosemicarbazide were used.

(0.27 g, 73%) obtained as orange crystals; mp 236–237 °C (from ethanol); IR (KBr): $v_{\rm max}/{\rm cm}^{-1}$ 3688 (N–H), 1628 (C=O), 1515 (C=N), 1122 (C=S); $\delta_{\rm H}$ (300 MHz; DMSO- d_6); 12.24 (1H, s, NNH), 9.14 (1H,

s, NHH), 8.87 (1H, s, NHH), 8.69 (1H, d, J 5.4, H-2"), 7.97 (1H, d, J 9.0, H-5"), 7.91 (1H, d, J 2.4, H-8"), 7.62 (1H, d, J, 2.1, H-4), 7.50 (1H, dd, J 2.1 and 7.8, H-6), 7.39 (1H, dd, J 2.4 and 9.0, H-6"), 7.35 (1H, d, J 7.8, H-7), 6.69 (1H, d, J 5.4, H-3"), 4.67 (2H, s, $-NCH_2N_{-}$), 3.15 (4H, br s, $-NCH_2CH_2NAr$), 2.89 (4H, br s, $-NCH_2CH_2N_-$); δ_C (75 MHz); 179.5, 161.8, 156.9, 152.1, 149.5, 143.7, 133.5, 131.6, 131.0, 130.9, 127.9, 126.1, 125.7, 122.9, 121.3, 119.3, 111.0, 109.4, 60.9, 52.5 (2 \times C) and 49.9 (2 \times C); HRMS (FAB) m/z558.04865 $[M+1]^{+}$ C₂₃H₂₁BrClN₆OS requires 558.04784. Anal. Calcd for C₂₃H₂₁BrClN₇OS: C 49.4%, H 3.8%, N 17.0%, S 5.7%. Found: C 50.2%, H 3.9%, N 17.2%, S 5.1%.

5.3.4. N¹-[4-(7-Chloro-quinolin-4-yl)-piperazin-1-ylmeth-yl-methyl-1*H*-indole-2,3-dione-3-thiosemicarbazone (2e). The conditions employed for the preparation of this compound were those described in general procedure **B**. However, 0.85 mmol compound **1e** and thiosemicarbazide were used.

(0.33 g, 78%) obtained as yellowish-orange crystals; mp 232–234 °C (from ethanol); IR (KBr): $v_{\text{max}}/\text{cm}^{-1}$ 3687 (N-H), 1675 (C=O), 1518 (C=N), 1122 (C=S); $\delta_{\rm H}$ (300 MHz; DMSO- $d_{\rm 6}$); 12.37 (1H, s, NNH), 9.17 (1H, s, NHH), 8.91 (1H, s, NHH), 8.68 (1H, d, J 5.4, H-2"), 7.99 (1H, d, J 9.0, H-5"), 7.94 (1H, d, J 2.1, H-8), 7.78 (1H, d, J 2.1, H-4), 7.55 (1H, dd, J 2.1 and 8.7, H-6), 7.48 (1H, d, J 2.1 and 9.0, H-6"), 7.34 (1H, d, J 8.7, H-7), 6.69 (1H, d, J 5.4, H-3"), 4.81 (2H, s, $-NCH_2N_-$), 3.16 (4H, br s, $-NCH_2CH_2NAr$), 2.90 (4H, br s, $-NCH_2CH_2NAr$), 2.27 (3H, s, Ar–C H_3); δ_C (75 MHz); 179.4, 161.7, 156.2, 152.8, 149.5, 144.0, 142.4, 134.6, 132.0, 130.8, 127.9, 126.1, 125.0, 123.6, 121.3, 120.0, 112.3, 109.4, 63.1, 51.6 $(2 \times C)$, 49.5 $(2 \times C)$ and 21.3; HRMS (FAB) m/z 494.15187 [M+1]⁺ C₂₄H₂₄ClN₇OS requires 494.15297. Anal. Calcd for C₂₄H₂₄ClN₇OS: C 58.4%, H 4.9%, N 19.8%, S 6.5%. Found: C 58.4%, H 4.9%, N19.7%, S 6.2%.

5.4. C. General procedure for preparation of compounds 5a-f

Equimolar quantities of isatin-3-thiosemicarbazone (0.20 g, 0.91 mmol), amine (0.09 mL, 0.91 mmol) and paraformaldehyde (0.08 g, 0.90 mmol) were dissolved in warm 99.9% ethanol. The mixture was continuously stirred at 25 °C for 6 h. After standing for approximately 24 h in the fridge, a crystalline product was formed. The product was then separated by filtration, vacuum dried and recrystallized from ethanol.

5.4.1. N^1 -Diethylaminomethyl-indole-2,3-dione-3-thiosemicarbazone (5a). (0.15 g, 54%) obtained as yellow crystals; mp 134–135 °C (ethanol) (lit. 135 °C);³³ δ_H (300 MHz; DMSO- d_6); 12.10 (1H, s, NNH), 9.05 (1H, s, NHH), 8.69 (1H, s, NHH), 7.71 (1H, dd, J 0.9 and 7.5, H-4), 7.44 (1H, ddd, J 0.9, 6.6 and 8.0, H-6), 7.17 (1H, ddd, J 0.9, 6.6 and 7.5, H-5), 7.19 (1H, d, J 8.0, H-7), 4.52 (2H, s, $-NCH_2N-$), 2.65 (4H, q, J 6.9, $-NCH_2CH_3$), 1.06 (6H, t, J 6.9, $-NCH_2CH_3$).

- **5.4.2.** N^{I} -Piperidin-1-ylmethyl-indole-2,3-dione-3-thiosemicarbazone (5b). (0.27 g, 95%); obtained as yellow solids; mp 179–180 °C (from ethanol) (lit. 174 °C);³³ δ_{H} (300 MHz, DMSO- d_6); 12.38 (1H, s, NNH), 9.05 (1H, s, N*H*H), 8.69 (1H,s, NH*H*), 7.71 (1H, dd, *J* 0.9 and 7.5, H-4), 7.41 (1H, ddd, *J* 0.9, 7.8 and 8.1, H-6), 7.26 (1H, d, *J* 7.8, H-7), 7.17 (1H, ddd, *J* 1.2, 7.5 and 8.1, H-5), 4.47 (2H, s, $-\text{NC}H_2\text{N}$ -), 2.54 (4H, t, *J* 4.8, H-2'), 1.46 (4H, m, H-3' 3'), 1.33 (2H, m, H-4' 4').
- **5.4.3.** N^1 -Pyrrolidin-1-ylmethyl-indole-2,3-dione-3-thiosemicarbazone (5c). (0.25 g, 90%) obtained as yellow crystals; mp 181–184 °C (from ethanol) (lit. 177–178 °C); 33 $\delta_{\rm H}$ (400 MHz, DMSO- d_6); 12.29 (1H, s, NNH), 9.03 (1H, s, N*H*H), 8.68 (1H, s, NH*H*), 7.68 (1H, dd, J 1.2 and 7.6, H-4), 7.42 (1H, ddd, J 1.2, 7.2 and 8.0, H-6), 7.23 (1H, d, J 8.0, H-7), 7.12 (1H, ddd, J 0.9, 7.2 and 7.6, H-5), 4.61 (2H, s, $-NCH_2N-$), 2.60 (4H, br t, J 3.2, H-2'), 1.65 (4H, m, H-3).
- **5.4.4.** N^1 -Morpholin-4-ylmethyl-indole-2,3-dione-3-thiosemicarbazone (5d). (0.27 g, 92%); as yellow crystals; mp 208–210 °C (from ethanol) (lit. 215–216 °C); ³³ $\delta_{\rm H}$ (300 MHz; DMSO- d_6); 12.37 (1H, s, NNH), 9.05 (1H, s, N*H*H), 8.70 (1H, s, NH*H*), 7.72 (1H, dd, *J* 0.6 and 7.2, H-4), 7.42 (1H, ddd, *J* 0.9, 7.5 and 8.0, H-6), 7.27 (1H, d, *J* 8.0, H-7), 7.15 (1H, ddd, *J* 0.6, 7.2 and 7.5, H-5), 4.49 (2H, s, $-NCH_2N-$), 3.54 (4H, br s, H-3'), 2.57 (4H, br s, H-2').
- 5.4.5. N^1 -4-Benzyl-piperazin-1-ylmethyl-indole-2,3-dione-**3-thiosemicarbazone** (5e). (0.23 g, 63%); obtained as yellow crystals; mp 187–188 °C; IR (KBr): $v_{\text{max}}/\text{cm}^{-1}$ 3680 (N-H), 3428 (N-H), 1637 (C=O), 1505 (C=N), 1123 (C=S); $\delta_{\rm H}$ (300 MHz, DMSO- d_6); 12.38 (1H, s, NNH), 9.06 (1H, s, NHH), 8.69 (1H, s, NHH), 7.71 (1H, dd, J 0.9 and 7.5, H-4), 7.43 (1H, ddd, J 0.9, 7.2 and 7.8, H-6), 7.25-7.23 (5H, m, H-2", H-3", H-4", H-5" and H-6"), 7.19 (1H, d, J 7.8, H-7), 7.17 (1H, ddd, J 0.6, 7.2 and 7.5, H-5), 4.52 (2H, s, $-NCH_2N_-$), 3.43 (2H, s, $-NCH_2Ar$), 2.59 (4H, br s, $-NCH_2CH_2NCH_2Ar$), 2.34 (4H, br s, $-NCH_2CH_2$ - NCH_2Ar); δ_C (75 MHz); 178.7, 161.6, 143.7, 138.2, 131.0, 130.9, 128.6 (2 \times C), 128.0, 126.8 (2 \times C), 122.8, 120.5, 119.2, 111.1, 61.8, 52.3 ($2 \times C$), 50.1 ($2 \times C$) and 49.9. Anal. Calcd for C₂₁H₂₄N₆OS: C 62.1%, H 6.6%, N 20.1%, S 6.6. Found: C 61.7%, H 5.9%, N 20.6%, S 6.6%; HRMS (FAB) m/z 409.18109 [M+1]⁺, C₂₁H₂₄N₆OS requires 409.18106.
- **5.4.6.** N^1 -**4-Phenyl-piperazin-1-ylmethyl-indole-2,3-dione-3-thiosemicarbazone (5f).** (0.27 g, 76%) obtained yellowish crystals; 188–191 °C; IR (KBr): $v_{\text{max}}/\text{cm}^{-1}$ 3686 (N–H), 3575 (N–H), 1694 (C=O), 1502 (C=N), 1127 (C=S); δ_{H} (300 MHz, DMSO- d_{6}); 12.39 (1H, s, NNH), 9.04 (1H, s, N*H*H), 8.69 (1H, s, NH*H*), 7.79 (1H, dd, *J* 0.9 and 7.6, H-4), 7.45 (1H, ddd, *J* 0.9, 6.6 and 8.0, H-6), 7.18 (1H, d, *J* 8.0, H-7), 7.21–7.13 (5H, m, H-2", H-3" and 4"), 7.10 (1H, ddd, *J* 0.6, 6.6 and 7.6, H-5), 4.59 (2H, s, $-\text{NC}H_{2}\text{N}-\text{N}$), 3.09 (4H, br s, $-\text{NC}H_{2}\text{C}H_{2}\text{NAr}$); δ_{C} (75 MHz); 178.7, 161.7, 151.0, 143.7, 131.1, 130.9, 128.8, 122.0, 120.6 (2×C), 120.5, 119.3 (2×C), 119.2,

111.1, 60.9, 50.1 (2 × C) and 48.1 (2 × C). Anal. Calcd for $C_{20}H_{22}N_6OS$: C 60.9%, H 5.6%, N 21.3%, S 8.1%. Found: C 61.3%, H 6.3%, N 18.7%, S 5.7%; HRMS (FAB) m/z 395.16544 [M+1]⁺, $C_{20}H_{22}N_6$ OS requires 395.16541.

5.5. D. General procedure for the preparation of compounds 4

Equimolar quantities of ethyl-quinoline isatin derivative (0.1 g, 0.284 mmol) and thiosemicarbazide (0.038 g, 0.284 mmol) were dissolved in warm 99.9% ethanol. The mixture was then continuously stirred at 45 °C for 5–6 h. After standing for approximately 24 h at room temperature, a crystalline product was formed. The product was then separated by filtration, vacuum dried and recrystallized from acetonitrile.

- N^1 -[2-(7-Chloro-quinolin-4-ylamino)-ethyl]-1Hindole-2,3-dione-3-thiosemicarbazone (4a). (0.08 g, 70%) obtained as orange crystals; mp 167-171 °C (from acetonitrile); IR (KBr): $v_{\text{max}}/\text{cm}^{-1}$ 3680 (N-H), 3404 (N-H), 3012 (Ar-H), 1635 (C=O), 1516 (C=N), 1140 (C=S); $\delta_{\rm H}$ (400 MHz, DMSO- d_6); 12.31 (1H, s, NNH), 9.02 (1H, s, NHH), 8.67 (1H, s, NHH), 8.40 (1H, d, J 5.4, H-2"), 7.98 (1H, d, J 9.0, H-5"), 7.76 (1H, d, J 2.0, H-8"), 7.67 (1H, dd, J 0.6 and 7.4, H-4), 7.38 (1H, dd, J 2.0 and 9.0, H-6"), 7.28 (1H, ddd, J 0.6, 7.6 and 7.8, H-6), 7.10 (1H, ddd, J 0.9, 7.4 and 7.6, H-5), 7.02 (1H, d, J 7.8, H-7), 6.63 (1H, d, J 5.4, H-3"), 4.01 (2H, t, J 5.6, -CH₂CH₂NHAr), 3.65 (2H, t, J 5.6, -CH₂CH₂NHAr); $\delta_{\rm C}$ (75 MHz); 179.4, 161.8, 152.6, 150.5, 149.7, 149.1, 143.7, 138.6, 134.1, 128.2, 124.9, 124.5, 123.7, 123.1, 118.2, 117.3, 110.3, 104.5, 74.9 and 42.4; HRMS (FAB) found m/z 425.09419 $[M+1]^+$, $C_{20}H_{17}ClN_6OS$] requires 425.09513. Anal. Calcd for C₂₀H₁₇ClN₆OS·0.5H₂O: C 55.4%, H 3.9%, N 18.8%, S 6.9%. Found: C 55.3%, H 4.0%, N 18.8%, S 6.9%.
- **5.5.2.** N^1 -[2-(7-Chloro-quinolin-4-ylamino)-ethyl]-5-methyl-1H-indole-2,3-dione-3-thiosemicarbazone (4b). The conditions employed for the preparation of this compound were those described in general procedure **D**. However, 0.27 mmol of compound 1b and thiosemicarbazide were used.
- (0.11 g, 89%) obtained as yellow-orange crystals; mp 227–230 °C (from ethanol); IR (KBr): $v_{\rm max}/{\rm cm}^{-1}$ 3632 (N–H), 3378 (N–H), 1665 (C=O), 1506 (C=N), 1127 (C=S); $\delta_{\rm H}$ (400 MHz; DMSO- $d_{\rm 6}$); 12.29 (1H, s, NNH), 8.98 (1H, s, NHH), 8.63 (1H, s, NHH), 8.40 (1H, d, J 5.6, H-2"), 8.01 (1H, d, J 9.0, H-5"), 7.76 (1H, d, J 2.0, H-8"), 7.49 (1H, d, J 2.0, H-4), 7.39 (1H, dd, J 2.0 and 9.0, H-6"), 7.08 (1H, dd, J 2.0 and 7.4, H-6), 7.00 (1H, d, J 7.4, H-7), 6.60 (1H, d, J 5.6, H-3"), 3.98 (2H, t, J 5.4, -NCH₂CH₂NHAr), 3.62 (2H, t, J 5.4, -NCH₂CH₂NHAr), 2.31 (3H, s, Ar-CH₃); $\delta_{\rm C}$ (75 MHz); 179.4, 161.8, 152.5, 150.6, 149.7 149.0, 141.6, 138.9, 134.2, 128.1, 125.3, 124.9, 124.5, 121.9, 118.2, 118.1, 110.9, 110.6, 74.9, 41.8 and 21.2; LRMS (FAB) m/z 439 [5%, (M+1)⁺], and 307 [30%, M-C₁₈H₁₄ClN₃]. Anal. Calcd for C₂₁H₁₉ClN₆OS: C

57.5%, H 4.4%, N 19.3%, S 7.3%. Found: C 57.1%, H 4.3%, N 18.8%, S 7.3%.

5.5.3. N^1 -[2-(7-Chloro-quinolin-4-ylamino)-ethyl]-5-chloro-1H-indole-2,3-dione 3-thiosemicarbazone (4c). The conditions employed for the preparation of this compound were those described in general procedure **D**. However, 0.26 mmol of compound 3c and thiosemicarbazide were used.

(0.16 g, 68%) obtained as orange crystals; mp 221-227 °C (from ethanol); IR (KBr): $v_{\text{max}}/\text{cm}^{-1}$ 3688 (N-H), 3478 (N-H), 1665 (C=O), 1510 (C=N), 1121 (C=S); $\delta_{\rm H}$ (400 MHz, DMSO- d_6); 12.29 (1H, s, NNH), 8.98 (1H, s, NHH), 8.64 (1H, s, NHH), 8.40 (1H, d, J 5.0, H-2"), 8.01 (1H, d, J 9.0, H-5"), 7.76 (1H, d, J 2.1, H-8"), 7.49 (1H, d, J 2.0, H-4), 7.39 (1H, dd, J 2.0 and 9.0, H-6"), 7.09 (1H, dd, J 2.0 and 7.4, H-6), 7.01 (1H, d, J 7.4, H-7), 6.60 (1H, d, J 5.0, H-3"), 3.98 (2H, t, J 6.4, -NCH₂CH₂NHAr), 3.62 (2H, t, J 6.4, $-NCH_2CH_2NHAr$); δ_C (75 MHz); 178.3, 158.2, 151.9, 150.8, 149.8, 149.0, 142.1, 136.7, 127.5, 127.3, 124.8, 124.2, 123.5, 121.9, 117.6, 117.0, 109.7, 104.7, 74.7 and 42.3; HRMS (FAB) found m/z 459.05659 [M+1]⁺, C₂₀H₁₆Cl₂N₆OS required 459.05615. Anal. Calcd for C₂₀H₁₆Cl₂N₆OS: C 57.5%, H 4.4%, N 19.3%, S 7.3%. Found: C 57.1%, H 4.3%, N 19.0%, S 7.2%.

5.5.4. 2-(7-Chloro-quinolin-4-ylamino)-ethanol (12). A mixture of 4,7-dichloroquinoline (4.0 g, 20.19 mmol), ethanolamine (24.42 mL, 403.9 mmol) and triethylamine (0.84 mL, 6.06 mmol) was stirred under reflux for 8 h. The reaction mixture was then cooled to room temperature during which a white precipitate formed. The precipitate was filtered, washed with water $(3 \times 50 \text{ mL})$ and recrystallized from methanol to obtain (4.10 g, 92%) as cream-white crystals, mp 213-215 °C (from methanol) (lit. 214 °C);¹⁹ R_f 0.42 (MeOH/DCM, 2:8); $\delta_{\rm H}$ (300 MHz; CD₃OD); 8.35 (1H, d, J 6.0, H-2'), 8.09 (1H, d, J 9.0, H-5'), 7.78 (1H, d, J 2.1, H-8'), 7.40 (1H, dd, J 2.1 and 9.0, H-6'), 6.57 (1H, d, J 6.0, H-3'), 3.86 (2H, t, J 5.7, ArNHCH₂CH₂-), 3.52 (2H, t, J 5.7, ArNHC H_2 -); δ_C (75 MHz); 151.8, 150.1, 149.0, 133.3, 127.4, 123.9, 117.4, 108.6, 58.7 and 45.0; LRMS (EI) m/z 223.1 [M⁺]. Anal. Calcd: C 59.3%, H 4.9%, N 12.6%. Found: C 59.0%, H 4.9%, N 12.6%.

5.5.5. Methanesulfonic acid 2-(7-chloro-quinolin-4-ylamino)-ethyl ester (13). A solution of compound 12 (8.0 g, 36 mmol) in 24 mL of pyridine was stirred at 0 °C for 30 min. Whilst at 0 °C methanesulfonyl chloride (6.97 mL, 90.0 mmol) in 22 mL of pyridine was added slowly to the above solution, and the resultant mixture stirred for 5 h. The reaction mixture was then diluted with 17% aqueous ammonia solution (25 mL), extracted with CH_2Cl_2 (3 × 50 mL), dried over anhydrous MgSO₄, concentrated and recrystallized from methanol/water to obtain yield (8.96 g, 83%) obtained as white crystals, mp 138–140 °C (from MeOH/ H_2O), R_f 0.76 (MeOH/DCM, 2:8); IR (KBr) $v_{\text{max}}/\text{cm}^{-1}$; 3218 (N-H), 1579 (C=N), 1442 (C=C); $\delta_{\rm H}$ (300 MHz; CD₃OD); 8.39 (1H, d, J 5.7, H-2'), 8.08 (1H, d, J 9.0, H-5'), 7.79 (1H, d, J 2.1, H-8'), 7.41 (1H, dd, J 2.1 and 9.0, H-6'), 6.61 (1H, d,

J 5.7, H-3′), 4.48 (2H, t, J 5.4, ArNHCH₂C H_2 –), 3.74 (2H, t, J, 5.4, ArNHC H_2 –) 3.07 (3H, s, –OSO₂–C H_3); $\delta_{\rm C}$ (100 MHz); 151.9, 150.2, 149.1, 133.2, 127.8, 124.1, 123.9, 117.9, 108.0, 66.8, 42.4 and 37.8; HRMS (EI) found m/z 300.03487 [M]⁺, C₁₂H₁₃O₃ClN₂S requires 300.03354. Anal. Calcd for C₁₂H₁₃O₃N₂SCl: C 47.9%, H 4.4%, N 9.3%, 10.7%. Found: C 48.1%, H 4.4%, N 9.2%, S 10.3%.

5.6. E. General procedure for preparation compounds 3a-e

To a mixture of isatin or substituted isatin (0.1 g, 0.68 mmol) in 10 mL of anhydrous DMF at 0 °C was added NaH (0.02 g, 0.815 mmol) was then added to this mixture followed by Compound 13 (0.27 g, 0.88 mmol) warming to room temperature with stirring over 1 h. The temperature was then raised to 60 °C and stirring continued for a further 12 h. Ice-cold water was added to the reaction and a coloured precipitate formed. The solid was filtered, washed with water (2 × 25 mL), petroleum ether (1 × 10 mL) and dried under vacuum to give the desired product.

 N^1 -[2-(7-Chloro-quinolin-4'-ylamino)-ethyl]-1H-5.6.1. indole-2,3-dione (3a). (0.21 g, 88%) obtained as orange crystals; mp 240–243 °C (from methanol); R_f 0.43 (MeOH/EtOAc, 2:8); IR (CHCl₃) $v_{\text{max}}/\text{cm}^{-1}$ 3236 (N– H), 1734 (keto C=O), 1600 (amide C=O), 1514 (C=N); $\delta_{\rm H}$ (300 MHz; DMSO- d_6); 8.41 (1H, d, J 5.4, H-2"), 7.97 (1H, d, J 9.0, H-5"), 7.76 (1H, d, J 2.4, H-8"), 7.52 (1H, dd, J 0.6 and 7.2, H-4), 7.48 (1H, ddd, J 0.6, 6.6 and 7.4, H-6), 7.39 (1H, dd, J 2.4, 9.0, H-6"), 7.07 (1H, ddd, J 0.9, 6.6 and 7.2, H-5), 7.03 (1H, dd, J 0.9 and 7.4, H-7), 6.62 (1H, d, J 5.4, H-3"), 3.93 (2H, t, J 5.1, ArNHCH₂CH₂–), 3.60 (2H, t, J 5.1, ArNHC H_2 -); δ_C (75 MHz); 183.3, 158.3, 151.9, 150.6, 149.7, 149.0, 137.8, 133.3, 127.4, 124.3, 124.0, 123.7, 122.9, 117.4, 117.0, 110.2, 104.6, 74.9 and 42.0; HRMS (FAB) found m/z 352.08494 [M+1]⁺, C₁₉H₁₄O₂N₃Cl] requires 352.08527. Anal. Calcd for C₁₉H₁₄O₂N₃Cl: C 64.8%, H 4.0%, N 11.9%. Found: C 64.4%, H 3.9%, N 11.2%.

5.6.2. N^1 -[2-(7-Chloro-quinolin-4-ylamino)-ethyl]-5-methyl-1H-indole-2,3-dione (3b). The conditions employed for the preparation of this compound were those described in general procedure E. However, 0.62 mmol of isatin and 0.81 mmol of compound 13 were used.

0.62 mmol (0.16 g, 68%) obtained as red crystals; mp 240–243 °C (from methanol); $R_{\rm f}$ 0.28 (MeOH/EtOAc; 2:8); IR (KBr) $v_{\rm max}/{\rm cm}^{-1}$ 3236 (N–H), 1737 (keto C=O), 1600 (amide C=O), 1514 (C=N); $\delta_{\rm H}$ (300 MHz; CDCl₃); 8.53 (1H, d, J 5.4, H-2"), 7.93 (1H, d, J 2.4, H-8"), 7.68 (1H, d, J 9.0, H-5"), 7.43 (1H, d, J 2.0, H-4), 7.39 (1H, dd, J 2.0 and 7.2, H-6), 7.37 (1H, dd, J 2.4 and 9.0, H-6"), 6.81 (1H, d, J 7.2, H-7), 6.62 (1H, d, J 5.4, H-3"), 4.15 (2H, t, J 5.4, ArNHCH₂CH₂-), 3.67 (2H, t, J 5.4, ArNHCH₂CH₂-), 3.67 (2H, t, J 5.4, ArNHCH₂-), 2.32 (3H, s, Ar-CH₃); $\delta_{\rm C}$ (75 MHz), 183.6, 158.2, 152.0, 150.5, 149.7, 149.1, 137.7, 133.4, 126.8, 124.6, 124.2, 123.4, 123.1, 117.5, 116.9, 111.2, 104.6, 74.6, 42.1 and 20.3; LRMS (EI) m/z 365.0 [16%, M⁺], 191

[100%, $M-C_{10}H_8ClN_2$]. Anal. Calcd for $C_{20}H_{16}O_2N_3Cl$: C 65.7%, H 4.4%, N 10.5%. Found: C 65.3%, H 4.3%, N 10.9%.

5.6.3. 5-Chloro- N^1 **-[2-(7-chloro-quinolin-4'-ylamino)-eth-yl]-1**H**-indole-2,3-dione (3c).** The conditions employed for the preparation of this compound were those described in general procedure E. However, 0.55 mmol of isatin and 0.72 mmol of compound **13** were used.

(0.15 g, 69%) obtained as orange crystals; mp 221-223 °C (from ethanol); R_f 0.46 (MeOH; EtOAc, 2:8); IR (CHCl₃) $v_{\text{max}}/\text{cm}^{-1}$, 3237 (N–H), 1734 (C=O), 1606 (amide C=O), 1519 (C=N); δ_H (300 MHz; DMSO- d_6); 8.43 (1H, d, J 5.4, H-2"), 7.94 (1H, d, J 8.4, H-5"), 7.78 (1H, d, J 2.1, H-8"), 7.59 (1H, d, J 2.4, H-4), 7.40 (1H, dd, J 2.4 and 8.7, H-6), 7.31 (1H, dd, J 2.1 and 8.4, H-6"), 7.07 (1H, d, J 8.7, H-7), 6.64 (1H, d, J 5.4, H-3"), 3.96 (2H, t, J 5.4, ArNHCH₂C H_2 -), 3.64 (2H, t, J 5.4, ArNHC H_2 -); δ_C (75 MHz); 182.3, 158.2, 151.9, 150.7, 149.8, 149.0, 136.7, 127.4, 127.2, 124.8, 124.2, 123.4, 121.9, 117.5, 117.0, 109.7, 104.7, 74.6 and HRMS (FAB) found m/z386.04642 $[M+1]^+C_{19}H_{13}O_2N_3Cl_2$ requires 386.04630. Anal. Calcd for C₁₉H₁₃O₂N₃Cl₂: C 59.1%, H 3.4%, N 10.9%. Found: C 59.0%, H 3.4%, N 10.8%.

5.6.4. *N*¹-[2-(7-Chloro-quinolin-4-ylamino)-ethyl]-5,7-dimethyl-1*H*-indole-2,3-dione (3d). The conditions employed for the preparation of this compound were those described in general procedure E. However, 0.57 mmol of isatin and 0.74 mmol of compound 13 were used.

(0.19 g, 89%) obtained as orange crystals; mp 219 °C (from ethanol); R_f 0.53; IR (CHCl₃) $v_{\text{max}}/\text{cm}^{-1}$ 3236 (N-H), 1736 (keto C=O), 1607 (amide C=O), 1519 (C=N); $\delta_{\rm H}$ (300 MHz; DMSO- d_6); 8.40 (1H, d, J 5.4, H-2"), 8.03 (1H, d, J 9.0, H-5"), 7.78 (1H, d, J 2.1, H-8"), 7.44 (1H, d, J 2.4, H-4), 7.40 (1H, d, J 2.4, H-6), 7.39 (1H, dd, J 2.1 and 9.0, H-6"), 6.62 (1H, d, J 5.4, H-3"), 4.12 (2H, t, J 6.3, ArNHCH₂CH₂-), 3.64 (2H, t, J 6.3, ArNHC H_2 -), 2.40 (3H, s, Ar-C H_3), 2.22 (3H, s, Ar–C H_3); δ_C (75 MHz); 182.2, 158.6, 151.8, 150.4, 149.8, 149.0, 136.3, 127.5, 127.4, 124.1, 123.7, 122.7, 121.2, 117.8, 117.4, 110.6, 104.6, 74.3, 42.7, 19.8 and 18.2; HRMS (FAB) found m/z 380.11717 [M+1] C₂₁H₁₈O₂N₃Cl requires 380.11657. Anal. Calcd for C₂₁H₁₈O₂N₃Cl: C 66.4%, H 4.7%, N 11.1%. Found: C 66.3%, H 4.6%, N 10.9%.

Acknowledgements

This material is based upon work supported by the South African National Research Foundation under grant number 2053362 (K.C.). We would also like to thank the University of Cape Town and the South African Medical Research Council for additional support. Work in the laboratories of P.J.R. was supported by grants from the National Institutes of Health (AI AI35800 and AI35707) and the Medicines for Malaria Venture. P.J.R. is a Doris Duke Charitable Foundation Distinguished Clinical Scientist.

References and notes

- World Health Organisation Fact Sheet No. 94. Revised Oct. WHO information. http://www.int/inf~fs/en/fact094. html 1998.
- Lima, P. D. C.; Barreiro, E. J. Avery, M. A. Abstracts of Papers, 223rd ACS National Meeting, Orlando, FL, USA, April 7–11, 2002; American Chemical Society: Washington, DC, 2002; MEDI-176.
- Flipo, M.; Florent, I.; Grellier, P.; Sergheraert, C.; Deprez-Poulain, R. Bioorg. Med. Chem. Lett. 2003, 13, 2659
- Romeo, S.; Dell'Agli, M.; Parapini, S.; Rizzi, L.; Galli, G.; Mondani, M.; Sparatore, A.; Taramelli, D.; Bosisio, E. Bioorg. Med. Chem. Lett. 2004, 14, 2931.
- Biot, C.; Dessolin, J.; Davioud-Charvet, E. Redox Rep. 2003, 8, 280.
- Davioud-Charvet, E.; Delarue, S.; Biot, C.; Schwobel, B.; Boehme, C. C.; Mussigbrodt, A.; Maes, L.; Sergheraert, C.; Grellier, P.; Schirmer, R. H.; Becker, K. J. Med. Chem. 2001, 44, 4268.
- Musonda, C. C.; Taylor, D.; Lehman, J.; Gut, J.; Rosenthal, P. J.; Chibale, K. Bioorg. Med. Chem. Lett. 2004, 14, 3901.
- 8. Lecaille, F.; Kaleta, J.; Bromme, D. Chem. Rev. 2002, 102, 4459
- Sajid, M.; McKerrow, J. H. Mol. Biochem. Parasitol. 2002, 120, 1.
- 10. Marquis, R. W. Ann. Rep. Med. Chem. 2000, 35, 309.
- Chiyanzu, I.; Hansell, E.; Gut, J.; Rosenthal, P. J.; McKerrow, J. H.; Chibale, K. Bioorg. Med. Chem. Lett. 2003, 13, 3527.
- Rosenthal, P. J.; Sijwali, P. S.; Singh, A.; Shenai, B. R. Curr. Pharm. Des. 2002, 8, 1659.
- 13. Egan, T. J. Drug Des. Rev.-Online 2004, 1, 93.
- 14. Biot, C. Curr. Med. Chem.-Anti-Infect. Agents 2004, 3, 135.
- Chibale, K.; Moss, J. R.; Blackie, M.; van Schalkwyk, D.; Smith, P. J. Tetrahedron Lett. 2000, 41, 6231.
- Ridley, R. G.; Hofheinz, W.; Matile, H.; Jaquet, C.; Dorn, A.; Masciadri, R.; Jolidon, S.; Richter, W. F.; Guenzi, A.; Girometta, M.-A.; Urwyler, H.; Huber, W.; Thaithong, S.; Peters, W. Antimicrob. Agents Chemother. 1996, 40, 1846
- 17. De, D.; Krogstad, F. M.; Byers, L. D.; Krogstad, D. J. *J. Med. Chem.* **1998**, *42*, 4918.
- Du, X.; Guo, C.; Hansell, E.; Doyle, P. S.; Caffrey, C. R.; Holler, T. P.; McKerrow, J. H.; Cohen, F. E. J. Med. Chem. 2002, 45, 2695.
- Egan, T. J.; Hunter, R.; Kaschula, C. H.; Marques, H. M.; Misplon, A.; Walden, J. J. Med. Chem. 2000, 43, 283
- Pradines, B.; Rolain, J. M.; Ramiandrasoa, F.; Fusai, T.; Mosnier, J.; Rogier, C.; Daries, W.; Baret, E.; Kunesch, G.; Le Bras, J.; Parzy, D. J. Antimicrob. Chemother. 2002, 50, 177.
- 21. Lundt, L.; Madsen, R. Synthesis 1992, 1129.
- 22. Semenov, A.; Olson, J. E.; Rosenthal, P. J. Antimicrob. Agents Chemother. 1998, 42, 2254.
- 23. Trager, W.; Jensen, J. Science 1976, 193, 673.
- 24. Makler, M. T.; Piper, R. C.; Milhous, W. K. *Parasitol. Today* **1998**, *14*, 376.
- Sijwali, P. S.; Kato, K.; Seydel, K. B.; Gut, J.; Lehman, J.; Klemba, M.; Goldberg, D. E.; Miller, L. H.; Rosenthal, P. J. *Proc. Natl. Acad. Sci. U.S.A.* **2004**, *101*, 8721.
- Greenbaum, D. C.; Baruch, A.; Grainger, M.; Bozdech, Z.; Medzihradszky, K. F.; Engel, J.; Derisi, J.; Holder, A. A.; Bogyo, M. Science 2002, 298, 2002.

- 27. Eksi, S.; Czesny, B.; Greenbaum, D. C.; Bogyo, M.; Williamson, K. C. *Mol. Microbiol.* **2004**, *53*, 243.
- 28. Greenbaum, D. C.; Mackey, Z.; Hansell, E.; Doyle, P.; Gut, J.; Caffrey, C. R.; Lehman, J.; Rosenthal, P. J.; McKerrow, J. H.; Chibale, K. J. Med. Chem. 2004, 47, 3212
- 29. Sweeney, D.; Raymer, M. L.; Lockwood, T. D. *Biochem. Pharmacol.* **2003**, *66*, 663.
- Eggleson, K. K.; Duffin, K. L.; Goldberg, D. E. J. Biol. Chem. 1999, 274, 411.
- 31. Moormann, A. M.; Hossler, P. A.; Meshnick, S. R. *Mol. Biochem. Parasitol.* **1999**, *98*, 283.
- 32. Fritsch, G.; Sawatzki, G.; Treumer, J.; Jung, A.; Spira, D. T. *Exp. Parasitol.* **1997**, *63*, 1.
- Rajendra, S. V.; Wobles, L. W. J. Med. Chem. 1967, 10, 972.